

# Effect of organic matter on porosity changes in thermally treated oil shale based on the differential $T_2$ ( $DT_2$ ) spectrum



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## ABSTRACT

Evaluating the porosity of oil shale at high temperatures is essential for resource exploration and development. However, the organic matter in oil shale rock, which is naturally rich in hydrogen, complicates assessments made using nuclear magnetic resonance (NMR) technology. In this paper, the differential  $T_2$  ( $DT_2$ ) spectrum method is introduced to correct the effects of organic matter on porosity measurements using nuclear magnetic resonance (NMR). Firstly, the  $T_2$  spectrum was obtained for both dry oil shale rock, which contains organic matter, and saturated oil shale rock, which contains both organic matter and pore water. Secondly, the  $DT_2$  spectrum of pore water in oil shale rock was obtained to determine the porosity of the oil shale rock containing organic matter. Finally, the impact of organic matter on the variation in porosity of oil shale rock as temperature increases was discussed. The results indicate that the porosity of oil shale rock increases slightly when the temperature is below 400 °C and rises significantly between 400 °C and 800 °C. The rate of porosity changes in oil shale rock increases gradually between 100 °C and 400 °C. The rate of porosity change in oil shale rock increases significantly and then decreases between 400 °C and 800 °C. There is a specific temperature at which the porosity changes of oil shale transitions from rapid to gradual.

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## 1. Introduction

The production of shale oil through the pyrolysis of organic matter at high temperatures is influenced by the porosity of the oil shale rock (Saif et al., 2017; Burnham, 2017; Rabbani et al., 2017; Alrwashdeh, 2022). Therefore, it is essential to investigate the variation in porosity of thermally treated oil shale rock (Kang et al., 2020; Yang et al., 2019; Wang et al., 2019).

Various experimental methods have been employed to investigate the variation in porosity of dry rock and rock masses at elevated temperatures (Fan et al., 2020; Gautam et al., 2018; Wu et al., 2020; Zheng et al., 2022). The variation in the microstructure of rock during the thermal process is typically attributed to two primary causes. The primary cause is thermal stress resulting from the inhomogeneous expansion of mineral constituents,

which leads to the initiation of micro-cracks when the stress exceeds the rock's strength (Rossi et al., 2018). The second cause is associated with mineral phase transitions in the rock, such as the transformation of  $\alpha$ -quartz to  $\beta$ -quartz at approximately 573 °C, which generates transgranular cracks and results in porosity variation in materials like granite (Feng et al., 2023; He et al., 2023; Yuan et al., 2022). To further explore the porosity changes in rock after thermal treatment, various laboratory technologies, such as X-ray computed tomography (CT) scans and gas adsorption, have been developed and frequently combined with digital image analysis methods (e.g., Digital Image Correlation or DIC) or other intelligent approaches to study the effects of high temperatures on porosity variations in rocks (Daigle and Johnson, 2016; Lei et al., 2021; Wang et al., 2021; Zhang et al., 2023).

In recent years, technologies have been developed to investigate the variation in porosity of rock and rock masses under saturated conditions at high temperatures, such as nuclear magnetic resonance (NMR) technology (Bi et al., 2023; Du et al., 2024). NMR technology measures the transverse relaxation time ( $T_2$ ) of hydrogen atoms in water, and this measurement can be utilized to characterize the

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volume of micro-defects in fully saturated rocks. NMR technology has been effectively utilized to study the porosity variations in thermally treated materials such as granite and sandstone, where only hydrogen atoms in pore water are present. However, for rocks that contain inherent hydrogen atoms, such as oil shale with organic matter, more complex methods are necessary. In these instances, two  $T_2$  spectra can be obtained—one from organic matter and the other from saturated water (Lei et al., 2021; Zhan et al., 2022). Thus, it is essential to develop a new NMR method, such as the differential  $T_2$  ( $DT_2$ ) spectrum method, to minimize the interference caused by organic matter. While the  $DT_2$  method is a known technique in NMR-based petrophysical analysis, its application to oil shale to correct hydrogen signal interference from organic matter is a significant and meaningful extension. It offers a unique perspective on understanding the effects of organic matter on porosity variation in thermally treated oil shale. Additionally, understanding the effects of organic matter on the variation of porosity in oil shale rock during or after exposure to high temperatures has significant implications for various applications.

In this study, the  $T_2$  spectrum was obtained for both dry oil shale rock containing organic matter and saturated oil shale rock that includes both organic matter and pore water. The  $DT_2$  spectrum of pore water in oil shale rock was obtained to determine the porosity of the oil shale rock containing organic matter. The impact of organic matter on the variation of porosity in oil shale rock as temperature increases was examined.

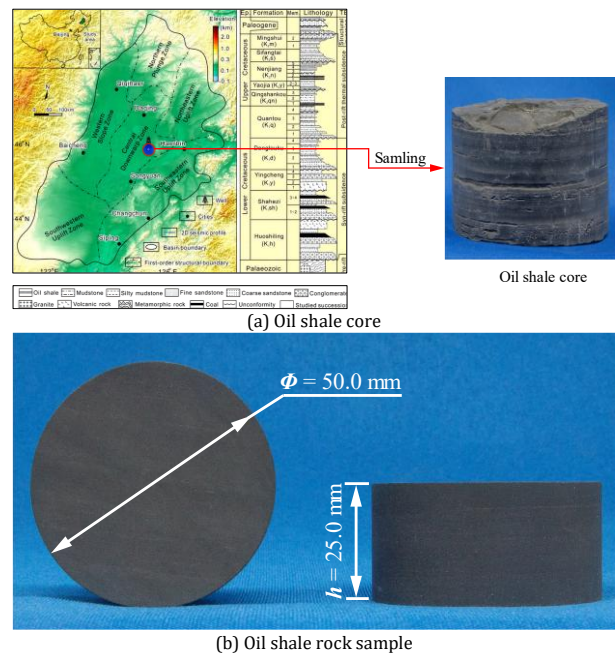
## 2. Samples and methods

### 2.1. Oil shale sample

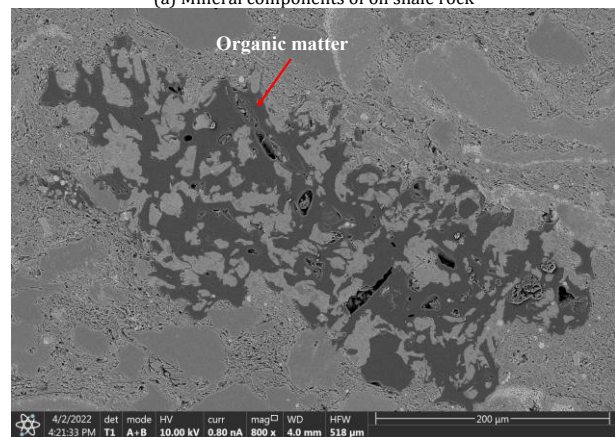
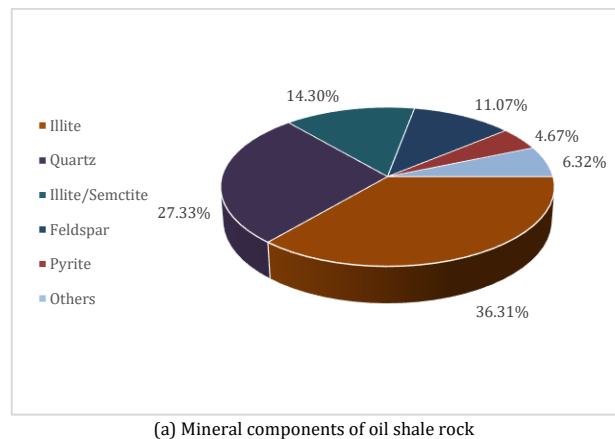
Fig. 1a illustrates the oil shale sample obtained from the Songliao Basin in China (Feng et al., 2010). The Songliao Basin is one of the most significant hydrocarbon-bearing basins in the country, renowned for its substantial oil and gas resources (Tian et al., 2024). Recent studies have highlighted the basin's potential for large-scale unconventional oil production, with estimates suggesting vast in-place resources of shale oil. The samples were collected from the Nenjiang Formation at a depth of 1,800 meters underground. These samples were subsequently processed into cylindrical shapes with a diameter of 50.0 mm and a length of 25.0 mm, as illustrated in Fig. 1b.

Fig. 2 shows the mineral composition and corresponding proportions of the oil shale rock, as determined by X-ray diffraction (XRD) and scanning electron microscopy (SEM). Fig. 2a illustrates that the oil shale is composed of illite (36.31%), quartz (27.33%), illite/smectite (14.30%), feldspar (11.07%), pyrite (4.67%), and other minerals (6.32%). Fig. 2b illustrates the morphology of organic matter, which is observed as black and clustered among inorganic minerals. The organic

matter in the oil shale from this region is sapropelic and demonstrates a high potential for hydrocarbon generation.



**Fig. 1:** Oil shale rock sample from the Songliao Basin, China (Feng et al., 2010)



**Fig. 2:** Components of oil shale rock

### 2.2. Experimental design

For a single-phase hydrogen-containing substance, the  $T_2$  spectrum obtained using NMR is the response characteristic of that phase of the

hydrogen-containing substance. But for rocks with multiphase hydrogen-containing substances, their signals of  $T_2$  spectrum overlap (Li et al., 2020; Ma et al., 2020). When the pore structure of rock is measured by NMR technology, the  $T_2$  spectrum of pore water is usually measured by filling with hydrogen-containing water. But oil shale itself contains hydrogen-rich organic matter. Therefore, it is necessary to use the differential  $T_2$  ( $DT_2$ ) spectrum method to distinguish pore water and organic matter.

Fig. 3 illustrates the experimental design for investigating the effects of organic matter on porosity variation in thermally treated oil shale, avoiding interference from hydrogen-rich organic matter. The experimental process can be divided into four stages: thermal treatment, the first NMR test, water-saturated treatment, and the second NMR test. Upon completing the experiments, the  $DT_2$  spectrum method was used to obtain the  $T_2$  spectra of pore water. It should be noted that in the calculation of the  $DT_2$  spectrum, the two  $T_2$  spectra are not directly calculated with a simple difference calculation. Specifically, the decay signals are first collected in thermal treatment and water-saturated state, the difference between the two decay signals is calculated, and then the  $T_2$  spectrum of pore water is obtained by inversion (Ge et al., 2017). In this paper, the analysis software of the equipment is used to calculate the  $DT_2$  spectrum. Then, the pore water  $T_2$  spectra were used to determine the pore size distribution and porosity after correcting for the effect of organic matter. The thermal effect on the microstructural deterioration of oil shale was analyzed by examining the porosity change rate.

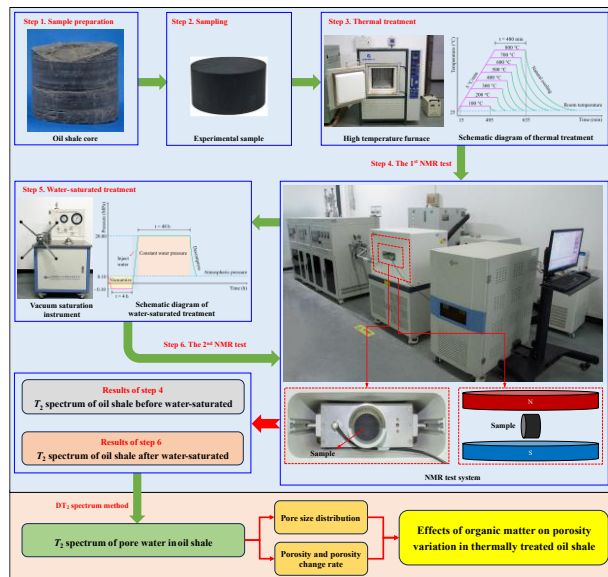


Fig. 3: Experimental design

### 2.2.1. Thermal treatment

Thermal treatments were performed using the SG-XL1200 high-temperature furnace. During the thermal treatments, a total of eight temperature levels were established: 100 °C, 200 °C, 300 °C, 400

°C, 500 °C, 600 °C, 700 °C, and 800 °C. Different samples were heated individually to their specified temperatures. The heating rate was maintained at 5.0 °C/min to prevent the effects of thermal shock (Saif et al., 2017; Fan et al., 2020). Upon reaching the specified temperatures, the samples were maintained for 8.0 hours before being cooled to room temperature.

### 2.2.2. NMR test

The NMR tests were conducted on samples using the Niumag MesoMR 14-060-I LF-NMR system. The resonance frequency was 12.00 MHz, and the diameter of the probe coil was 60.00 mm. The Carr-Purcell-Meiboom-Gill (CPMG) sequence was employed to evaluate the  $T_2$  spectrum, using an echo time of 0.10 ms and a waiting time of 3000.00 ms (Li et al., 2020). Furthermore, the Half-Fourier acquisition single-shot turbo spin echo imaging sequence was utilized for nuclear magnetic resonance imaging (MRI) testing, featuring a repetition time of 300.00 ms and an echo time of 4.53 ms.

### 2.2.3. Water-saturated treatment

Water-saturated treatments were conducted on thermally treated samples using the ZYB-II vacuum pressurization saturation device. Distilled water was used for saturation. During the water-saturated treatments, the oil shale rocks were initially subjected to vacuum for 4.0 hours in an environment with a pressure of -0.10 MPa. They were then submerged in water for 48.0 hours at a pressure of 20.00 MPa.

## 3. Results

### 3.1. Oil shale rock following thermal treatment

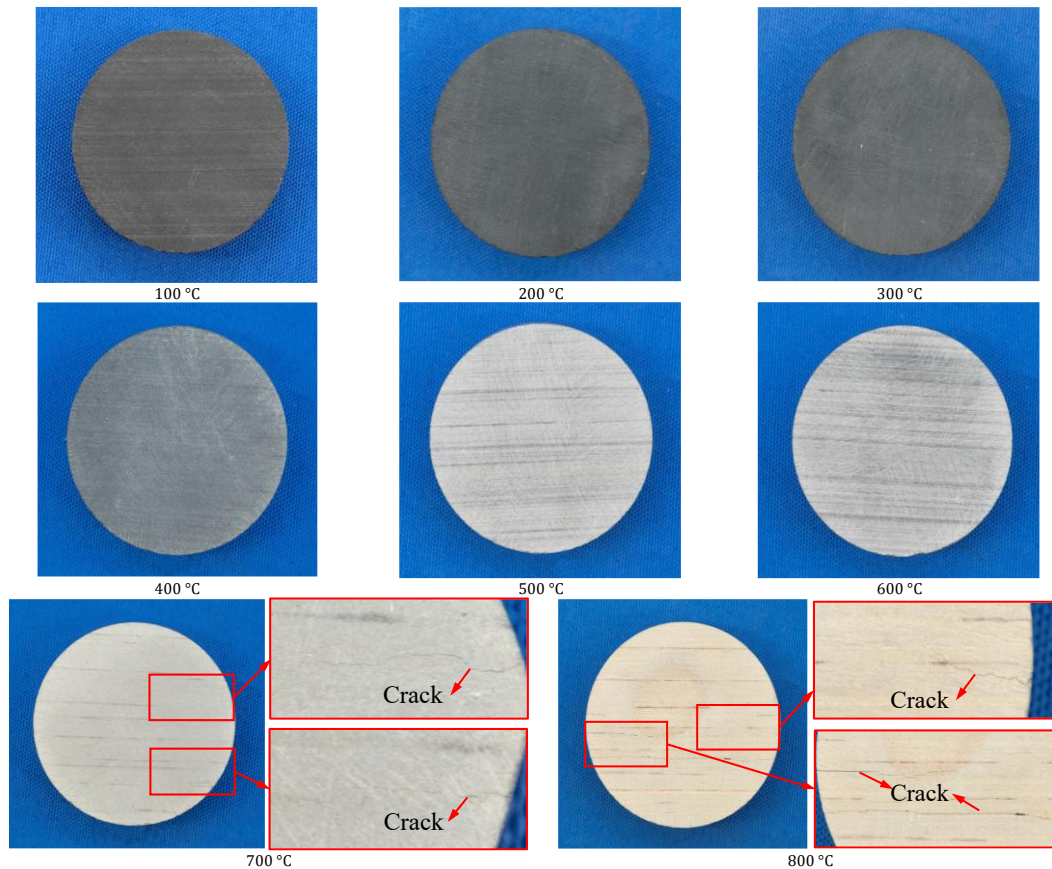
Fig. 4 illustrates the variation in the appearance of oil shale rock at different temperatures. The apparent color of oil shale remains predominantly black at temperatures 100 °C to 300 °C. Subsequently, it transitions from black to dark gray between 300 °C and 400 °C. Further heating results in a gradual transition in apparent color from dark gray to light gray between 400 °C and 600 °C. Finally, at 700 °C to 800 °C, the apparent color changes from light gray to pale yellow. Additionally, elevated temperatures lead to the formation of thermal cracks, including prominent surface cracks observed at 700 °C and 800 °C. Moreover, the observed crack propagation is aligned parallel to the bedding direction.

### 3.2. MRI of saturated oil shale rock

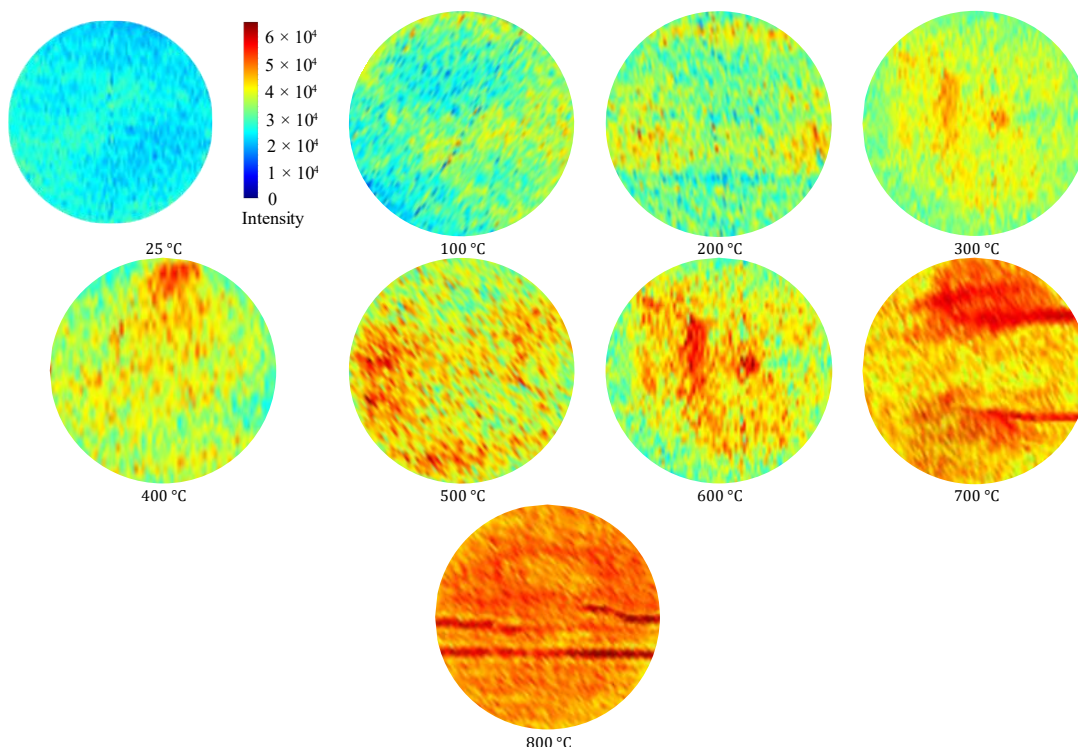
Fig. 5 illustrates the results of the MRI conducted at various temperatures on a cross-sectional scan area that encompasses the entire rock under water-

saturated conditions, where hydrogen signals originate from both organic matter and pore water. The red area gradually expands, indicating a progressive increase in the signal intensity of hydrogen atoms. This indicates an augmentation in both water content and pore quantity within the oil shale. The presence of ribbon-like red regions was observed at 700 °C and 800 °C. The hydrogen signal

in this region is attributed to the presence of water in the cracks, with a higher concentration of water being detected. To summarize, the MRI results indicate that high temperature leads to the formation of pores in oil shale rock and promotes the expansion of fractures from a macroscopic perspective.



**Fig. 4:** Appearance variation of oil shale rock after thermal treatment



**Fig. 5:** Oil shale rock after thermal treatment

### 3.3. T<sub>2</sub> spectrum of organic matter and pore water in oil shale rock

Fig. 6 illustrates the T<sub>2</sub> spectrum of organic matter in oil shale rock. The T<sub>2</sub> spectrum exhibits two peaks at 25 °C, while at 100 °C, it displays only one peak. Additionally, there is a decrease in the T<sub>2</sub> spectrum area from  $6.46 \times 10^3$  a.u. The value decreased to  $4.08 \times 10^3$  a.u., representing a decrease of 36.81%. This decrease in hydrogen signal intensity observed in oil shale at this temperature may be attributed to the volatilization of water and oil present within the rock matrix. The curve subsequently displays an additional downward shift between 100 °C and 200 °C. The T<sub>2</sub> spectrum area experiences a decrease from  $4.08 \times 10^3$  a.u. to  $3.37 \times 10^3$  a.u. with a decrease of 17.47%. The curve remains relatively stable between 200 °C and 400 °C, and the T<sub>2</sub> spectrum area exhibits a decrease from  $3.37 \times 10^3$  a.u. to  $3.06 \times 10^3$  a.u., which corresponds to a decrease of 9.24%. Subsequently, at 400 °C and 500 °C, the curve shifts downwards, and the T<sub>2</sub> spectrum area decreases significantly from  $3.06 \times 10^3$  a.u. to  $1.41 \times 10^3$  a.u., indicating a substantial reduction of approximately 53.79%. This change further indicates that organic matter in oil shale rock may be significantly decomposed between 400 °C and 500 °C. Moreover, at temperatures of 500 °C and 600 °C, there is a further decrease in the T<sub>2</sub> spectrum area from  $1.41 \times 10^3$  a.u. to  $0.83 \times 10^3$  a.u. The percentage drop of 41.19%. Notably, when temperature is 600 °C, 700 °C, and 800 °C, the T<sub>2</sub> spectrum areas are measured as  $0.62 \times 10^3$  a.u.,  $0.37 \times 10^3$  a.u., and  $0.2 \times 10^3$  a.u. with a further decrease.

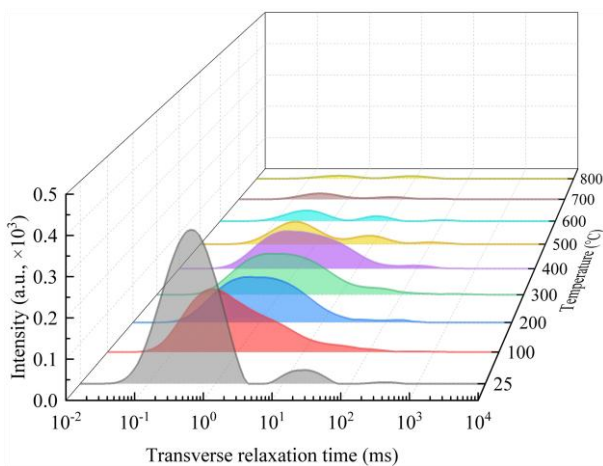


Fig. 6: T<sub>2</sub> spectra of organic matter

Fig. 7 illustrates the T<sub>2</sub> spectrum of organic matter and pore water in oil shale rock. At 25 °C to 100 °C, the area of the T<sub>2</sub> spectrum increases from  $1.61 \times 10^3$  a.u. to  $1.97 \times 10^3$  a.u. with an increase of 22.58%. At temperatures ranging from 200 °C to 400 °C, the T<sub>2</sub> spectrum area gradually increases, and it is  $2.08 \times 10^3$  a.u. (200 °C),  $2.25 \times 10^3$  a.u. (300 °C), and  $2.26 \times 10^3$  a.u. (400 °C), respectively. At 500 °C, the T<sub>2</sub> spectrum area increases to  $2.56 \times 10^3$  a.u. with an increase of 12.89% compared with 400 °C. The T<sub>2</sub> spectrum area increases from  $2.56 \times 10^3$  a.u. to  $2.86 \times 10^3$  a.u.

$\times 10^3$  a.u. at temperatures ranging from 500 °C to 600 °C, an increase of 12.20%. At 700 °C, the area of the T<sub>2</sub> spectrum increases obviously to  $4.69 \times 10^3$  a.u., an increase of 64.06%. At 800 °C, the T<sub>2</sub> spectrum area continues to increase, reaching  $5.08 \times 10^3$  a.u.. The above experimental phenomena show that after thermal treatment, the hydrogen signal of hydrogen-containing substances in oil shale under water-saturated states increases with the temperature increase, and this phenomenon is more significant between 600 °C and 700 °C.

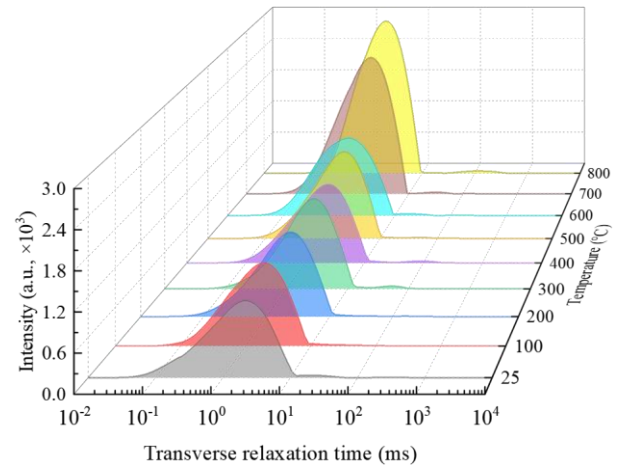


Fig. 7: T<sub>2</sub> spectra of organic matter and pore water

### 3.4. DT<sub>2</sub> spectrum of pore water in oil shale rock

Fig. 8 presents the DT<sub>2</sub> spectrum of pore water in oil shale rock, derived from the difference between Fig. 6 and Fig. 7. The T<sub>2</sub> spectrum area increases from  $1.97 \times 10^3$  a.u. to  $2.10 \times 10^3$  a.u. at 100 °C to 200 °C, there is an increase of 6.66%. The T<sub>2</sub> spectrum area increases from  $2.10 \times 10^3$  a.u. to  $2.25 \times 10^3$  a.u. at 200 °C to 300 °C, there is an increase of 6.93%. Between 300 °C and 400 °C, the curve shifts upwards, and the T<sub>2</sub> spectrum area increases from  $2.25 \times 10^3$  a.u. to  $2.26 \times 10^3$  a.u. with an increase of 0.53%. Between 400 °C and 500 °C, the curve shifts upwards and the T<sub>2</sub> spectrum area increases from  $2.26 \times 10^3$  a.u. to  $2.65 \times 10^3$  a.u. with an increase of 17.51%. Between 500 °C and 600 °C, the T<sub>2</sub> spectrum area increases from  $2.65 \times 10^3$  a.u. to  $2.86 \times 10^3$  a.u. with an increase of 7.75%. Between 600 °C and 700 °C, the curve shifts sharply upwards and the T<sub>2</sub> spectrum area increases from  $2.86 \times 10^3$  a.u. to  $4.69 \times 10^3$  a.u. with an increase of 64.01%. Finally, the curve continues to shift upwards between 700 °C and 800 °C, the T<sub>2</sub> spectrum area increases from  $4.69 \times 10^3$  a.u. to  $5.06 \times 10^3$  a.u. with an increase of 7.81%.

### 3.5. Pore distribution of thermally treated oil shale rock

Transverse relaxation time (T<sub>2</sub>) under a gradient magnetic field mainly includes surface relaxation (T<sub>2s</sub>), bulk fluid relaxation (T<sub>2B</sub>), and diffusion relaxation (T<sub>2D</sub>), and is expressed as

$$\frac{1}{T_2} = \frac{1}{T_{2s}} + \frac{1}{T_{2B}} + \frac{1}{T_{2D}} \quad (1)$$

However, the low-field NMR relaxation measurements  $T_2$  are primarily determined by the surface relaxation (Du et al., 2024; Zhang et al., 2019). Therefore,  $T_2$  can be expressed as

$$\frac{1}{T_2} = \rho_2 \frac{S}{V} = \rho_2 \frac{a}{r} \quad (2)$$

where,  $S/V$  is the surface-to-volume ratio of the pore ( $\mu\text{m}^{-1}$ ),  $a$  is the pore shape constant (2 for cylindrical pores and 3 for spherical pores),  $\rho_2$  is surface relaxivity ( $\mu\text{m}/\text{ms}$ ), a material-specific parameter quantifying the relaxation efficiency at the pore-fluid interface. In organic-rich shale,  $\rho_2$  is primarily influenced by the interfaces between clay and organic matter, which show minimal variation under static saturation conditions. Prior studies report  $\rho_2$  ranges of 0.003 - 0.010  $\mu\text{m}/\text{ms}$  for shales, influenced by organic content. Based on previous research recommendations, this paper adopts a surface relaxivity value of  $\rho_2 = 0.007 \mu\text{m}/\text{ms}$  (Zhang et al., 2019). However, it should be noted that the selection of  $\rho_2$  has an impact on the calculated pore radius.

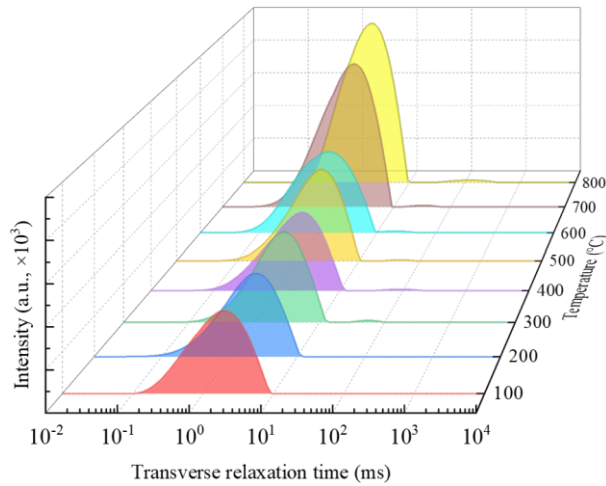


Fig. 8: DT<sub>2</sub> spectra of pore water

Based on the DT<sub>2</sub> spectra of pore water in oil shale presented in Fig. 8, the pore distribution of the oil shale rock can be determined, as illustrated in Fig. 9. Fig. 9 illustrates that the minimum pore radius of oil shale rock is 1.17 nm at 100 °C. Additionally, when the pore radius is 19.17 nm, the porosity component is the maximum (0.25%). It indicates that oil shale has a nanoscale pore structure. Furthermore, the maximum porosity components are observed at temperatures of 200 °C (0.30%), 300 °C (0.34%), 400 °C (0.30%), and 500 °C (0.36%) with corresponding pore radius measuring at 22.04 nm, 29.14 nm, 25.34 nm, and 22.04 nm, respectively. The left wing of the pore distribution curve clearly shifts upward, obviously between 500 °C and 600 °C. It indicates that the porosity of smaller pores varies significantly. At 600 °C and 700 °C, the curve shifts sharply upward, and the maximum porosity component increases from 0.34% to 0.63%. Moreover, the curve demonstrates a continuous upward shift, and the maximum porosity component increases from 0.63% to 0.72% between 700 °C and

800 °C. Additionally, the minimum pore radius of oil shale rock is measured at 1.56 nm at 800 °C, which surpasses that detectable at 700 °C (0.14 nm). This illustrates that not only does the number of pores increase, but also the smaller pores merge into larger ones in shale rock oil at high temperature.

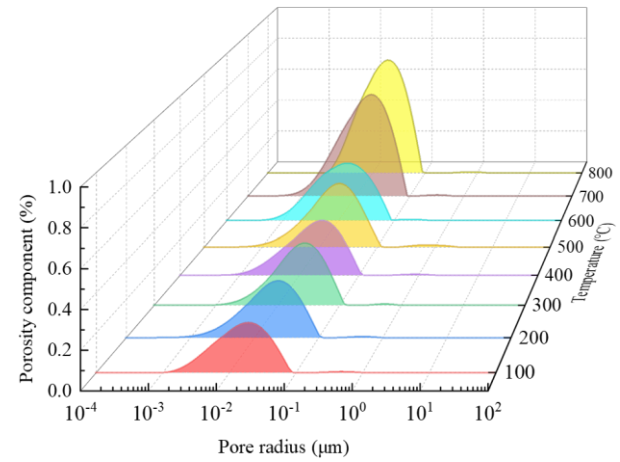


Fig. 9: Pore distribution of oil shale rock based on the DT<sub>2</sub> spectra of pore water

## 4. Discussion

### 4.1. Effects of temperature on porosity variation

Porosity is defined as the ratio of pore volume to the total volume of the sample. Pore volume can be determined from the DT<sub>2</sub> spectrum of pore water, which corresponds to the volume of the pore water itself. Fig. 10 illustrates the variation in porosity of oil shale rock as temperature changes. The increase in porosity ( $\phi$ ) with temperature ( $T$ ) was modeled using a modified Boltzmann sigmoidal function as

$$\phi(T) = \frac{\phi_{\max} - \phi_{\min}}{1 + e^{-\frac{T - T_c}{k}}} + \phi_{\min} \quad (3)$$

where,  $\phi_{\min}$  and  $\phi_{\max}$  denote the initial porosity at ambient temperature and maximum porosity, respectively.  $T_c$  is the critical inflection temperature, and  $k$  is the function parameter.

In the present study,  $\phi_{\min}$  and  $\phi_{\max}$  is fitted as 4.16% and 9.99%,  $T_c$  and  $k$  are determined as 658.0 °C and 85.0 °C based on the experimental test, respectively. Fig. 10 shows that this model demonstrated excellent agreement with experimental data ( $R^2 = 0.99$ ), which effectively captures the dual-phase evolution of pore development in oil shale. The porosity of oil shale rock increases slightly from 4.16% to 4.88% when subjected to temperatures between 100 °C and 400 °C, representing an increase of 17.31%. The porosity of oil shale rock increases significantly between 400 °C and 800 °C.

At 400 °C to 600 °C, the porosity of oil shale rock increases from 4.88% to 6.16%, representing a rise of 26.23%. At 600 °C to 800 °C, the porosity of oil shale rock increases from 6.16% to 9.99%, representing an increase of 62.18%.

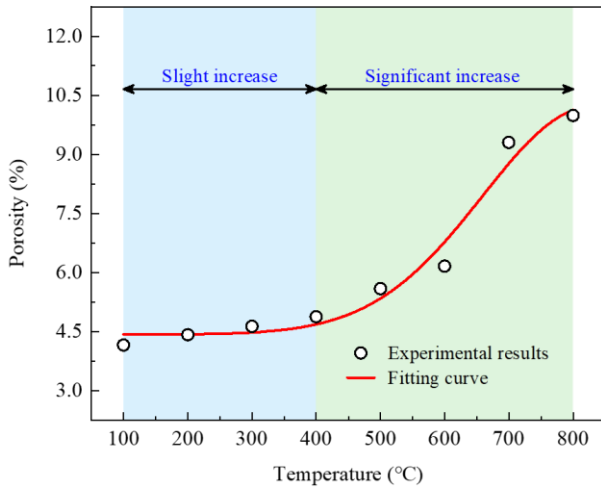


Fig. 10: Porosity variation as temperature increases

#### 4.2. Effects of temperature on variation of porosity change rate

This paper introduces the porosity change rate to further examine the effects of high temperature on the microstructural deterioration of oil shale rock. This rate represents the variation in porosity per unit temperature, as shown in Fig. 11. The porosity change rate of oil shale rock gradually increases slowly from  $0.02 \times 10^{-3} \text{ \%/}^\circ\text{C}$  to  $3.36 \times 10^{-3} \text{ \%/}^\circ\text{C}$  at 100 °C to 400 °C, and further indicates that the deterioration of oil shale rock is slow. As the temperature increases to approximately 658 °C, the porosity change rate is the maximum ( $20.50 \times 10^{-3} \text{ \%/}^\circ\text{C}$ ), and it is 5.10 times higher than that at 400 °C, further indicating that the deterioration of oil shale rock is accelerated. However, the deterioration of oil shale rock does not continuously increase, especially between 658 °C and 800 °C; the deterioration of oil shale rock is step-down. The above phenomenon indicates that, in addition to the effects of organic matter on the porosity (between 100 °C and 600 °C), the thermal crack is still the dominant cause for the porosity variation (between 600 °C and 800 °C).

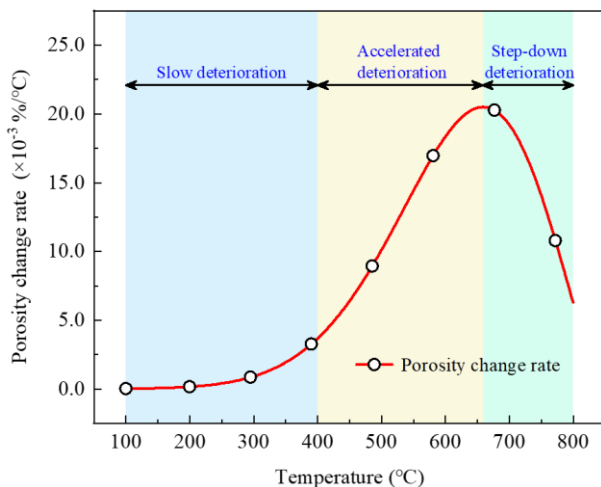


Fig. 11: Porosity change rate variation as temperature increases

According to the results of pore size distribution, porosity and porosity change rate of oil shale at high

temperature, the pore structure evolution is governed by the synergistic effects of organic pyrolysis, thermal stress, and mineral phase transitions. Between 100 °C and 400 °C, gradual organic matter pyrolysis generates hydrocarbon products and nanoscale pores (minimum radius: 1.17 nm - 1.56 nm), driving a moderate porosity increase from 4.16% to 4.88% and shifting peak porosity components to larger radii (19.17 nm - 29.14 nm). Between 400 °C and 600 °C, intensified thermal cracking accelerates organic decomposition, expanding pore networks (porosity rises 4.88% to 6.16%) and inducing upward shifts in small-pore distribution curves (between 500 °C and 600 °C), while peak porosity components fluctuate (22.04 nm- 29.14 nm) due to dynamic pore coalescence. Above 600 °C, thermal stress becomes the dominant factor, leading to the propagation of microcracks. This process significantly increases porosity by 62.18% (6.16% to 9.99%) and merges smaller pores (minimum radius increases from 0.14 nm at 700 °C to 1.56 nm at 800 °C). The rate of porosity change peaks at 658 °C ( $20.50 \times 10^{-3} \text{ \%/}^\circ\text{C}$ ), marking the transition from organic-driven to fracture-dominated mechanisms.

Based on the research results of this study, the significant porosity increase observed at > 400 °C aligns with the onset of kerogen pyrolysis during underground heating. The DT<sub>2</sub>-derived porosity model can analyze pore network development during in-situ conversion processes (ICP), informing optimal heating zone design (e.g., well spacing and target temperature profiles).

The upward shift in pore distribution curves at > 500 °C indicates that pre-heating shale reservoirs to 400 °C - 600 °C could enhance nano-pore connectivity, potentially reducing the fracturing pressure required to initiate cracks. MRI results confirm that thermal cracks propagate parallel to bedding planes. Field operations could leverage this anisotropy by aligning hydraulic fractures with natural thermal crack networks to maximize stimulated reservoir volume.

#### 5. Conclusions

This study investigated the thermal effects on the porosity variation of naturally hydrogen-rich oil shale rock using the DT<sub>2</sub> method, which was employed to differentiate between organic matter and pore water. The conclusions are as follows:

1. The organic matter and pore water in oil shale rock were distinguished by the DT<sub>2</sub> spectrum, which provides a new approach to correct hydrogen signal interference from organic matter, which is a common issue in NMR analysis of oil shale. Based on the DT<sub>2</sub> spectrum, the workflow designed in this paper quantifies organic matter contribution to porosity under high temperature conditions.
2. The porosity of oil shale rock increases slightly when the temperature is below 400 °C. The

porosity of oil shale rock increases significantly between 400 °C and 800 °C. Besides the effects of organic matter on porosity at temperatures ranging from 100 °C to 600 °C, thermal cracking remains the dominant cause of porosity variation between 600 °C and 800 °C.

3. The rate of porosity changes in oil shale rock increases gradually between 100 °C and 400 °C. The temperature increases significantly and then decreases between 400 °C and 800 °C. The most significant effect of high temperature on the porosity of oil shale rock occurs at a temperature of approximate 658 °C in the present study, where the rate of porosity change reaches its maximum.
4. The temperature-dependent porosity thresholds (400 °C and 658 °C in the present study) and DT<sub>2</sub> methodology provide actionable parameters for optimizing in-situ conversion, hydraulic fracturing, and log interpretation in organic-rich shale reservoirs.

## Compliance with ethical standards

## Conflict of interest

The author(s) declared no potential conflicts of interest with respect to the research, authorship, and/or publication of this article.

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